(-)-MAGLIFLOENONE, A NOVEL SPIROCYCLOHEXADIENONE NEOLIGNAN AND OTHER CONSTITUENTS FROM MAGNOLIA LILIFLORA

BANI TALAPATRA, PRABIR K. CHAUDHURI and SUNIL K. TALAPATRA

Department of Chemistry, University College of Science, Calcutta 700009, India

(Revised received 11 August 1981)

Key Word Index—Magnolia liliflora; Magnoliaceae; tetrahydrofuranolignan; (+)-veraguensin; neolignans; futoenone; (-)-maglifloenone; aporphine derived alkaloid; taspine; β -sitosterol; mass spectral study.

Abstract—A new neolignan designated (-)-maglifloenone and the known one futoenone, both of which contain the rarely occurring spirocyclohexadienone skeleton, have been isolated together with the tetrahydro-furanolignan (+)-veraguensin, an optically inactive tertiary base taspine and β -sitosterol from the leaves and twigs of *Magnolia liliflora*. The structure and stereochemistry of (-)-maglifloenone have been deduced from the spectral data and the mass fragmentation of (-)-maglifloenone and futoenone have been rationalized. This is the first report of two neolignans of spirocyclohexadienone skeleton and of taspine from the Magnoliaceae family and the second report of the natural occurrence of futoenone.

INTRODUCTION

Magnolia lilistora Desr., a large handsome tree native to Japan was collected in June, 1979 in Darjeeling where it has been naturalized. The wood of this species is used for lacquer ware in Japan [1]. The ether extract of M. lilistora depressed the spontaneous activity of mice and chicks and caused a distinct muscle weakness. In addition it suppressed convulsions induced by strychnine, picrotoxin and pentetrazole [2]. The bark of M. lilistora was found to contain two neolignans, magnolol and honokitol [3], and an alkaloid magnocuranine [1] whereas the leaves and roots were found to contain aporphine, benzylisoquinoline alkaloids [4] and quarternary alkaloids [5-7].

RESULTS AND DISCUSSION

Extensive chromatography over Si gel of the petrol and chloroform extracts of the dried and milled leaves and twigs of *M. liliflora* led to the isolation of a new neolignan with a spirocyclohexadienone

skeleton which we have designated (-)-maglifloenone (1), in addition to four other known compounds: futoenone (2), the only reported neolignan possessing a spirocyclohexadienone skeleton, (+)-veraguensin (3), a 2,5-diaryltetrahydrofuranolignan [8], the optically inactive t-base taspine (4) and β -sitosterol. To our knowledge, the present paper constitutes the first report of the alkaloid taspine and the above lignans except (+)-veraguensin [9], in the Magnoliaceae family and the second report of the natural occurrence of futoenone.

(-)-Maglifloenone, mp 232°, $C_{22}H_{26}O_6$ (M⁺ 386), $[\alpha]_D = -90.12^\circ$ (CHCl₃; c 0.081) has been assigned structure 1 from spectral evidence. Its IR spectrum (KBr) shows a conjugated carbonyl band at 1658 cm⁻¹ but no hydroxyl absorption. A comparison of the ¹H NMR spectral data (Table 1) of (-)-maglifloenone with those obtained by us for the congener neolignan futoenone (2) [10–12], mp 192°, $[\alpha]_D = -60^\circ$ (CHCl₃) leads to the structure and stereochemistry depicted in 1. The difference in

1 R' = R'' = R''' = OMc
2 R' = R'' =
$$CH_2$$
 O^- , R''' = H

aromatic substitution patterns between 1 and 2 is clearly evident from Table 1. (-)-Maglifloenone (1) exhibits three aromatic methoxyl signals and a 2H singlet assignable to H-2 and H-6 whereas futoenone (2) shows one methylenedioxy singlet (2H) and a

broad 3H signal due to H-2, H-5 and H-6. The signals due to the remaining 15 protons are almost identical in the spectra of 1 and 2. Futoenone (2) has only been isolated from one other plant, *Piper futokadzura* Sieb et Zucc (Fam. Piperaceae) [10].

Scheme 1. Mass fragmentation of maglifloenone (1) and futoenone (2). In each case the ion with the higher m/z value is from 1 and the one with the lower m/z value is from 2.

Table 1. ¹H NMR spectral data of compounds 1 and 2 [80 MHz (1) and 60 MHz (2), CDCl₃, TMS as int. standard]

Assignment	1	2
<u>Н</u> ₃ С-8	0.63 d	0.62 d
	$J_{9.8} = 6.4$	$J_{9.8} = 6.4$
H-7, H-8, H-7' and H-9'	$1.5-3.0 \ m(br)$	$1.5-3.0 \ m(br)$
H₃CO-5′	3.67 s	3.70 s
H-8'	5.04 m	5.05 m
H-6'	5.47 s	5.47 s
H-3'	5.78 s	5.80 s
CH ₂ O ₂ at C-3 and C-6		5.96 s
H ₃ CO-4	3.82 s	_
H ₃ CO-3 and H ₃ CO-5	3.85 s	_
H-2 and H-6	6.37 s	
H-2, H-5 and H-6		$6.63-6.80 \ s(br)$

The UV absorption of 1 [λ_{\max}^{EIOH} nm (log ϵ): 258 (4.32) and 285 (inflection)] is also in accord with the presence of an α -methoxy- α , β , α' , β' -dienone chromophore (calc. 262) [13].

The mass spectra of maglifloenone (1) and futoenone (2) are characteristic of the spirodienone skeleton and the genesis of the significant ions in both cases is given in Scheme 1. The four alternative pathways of the fission of the saturated spiro-ring B is noteworthy. In the mass spectrum of futoenone (2) (the genesis of its mass ions has not been reported earlier) the ions containing ring A are the same and those containing ring C appeared with the expected mass shifts (46 a.m.u. less than 1).

The third lignan, mp 125°, $[\alpha]_D = +31.8^\circ$, had similar UV, IR, 'H NMR and MS spectral data as those reported for (+)-veraguensin (3) isolated from Ocotea veraguensis Mez. (Fam. Lauraceae) [14]. The basic fraction of the chloroform extract of M. liliflora upon usual work-up afforded taspine (4), mp 225°d, hydrochloride 251-252°d, $[\alpha]_D = \pm 0^\circ$, isolated earlier from two Berberidaceae plants Leontice eversmannii Bunge and Caulophyllum robustum Maxim [15, 16].

EXPERIMENTAL

Mps are uncorr. Si gel (100-200 mesh) was used for chromatography, unless otherwise stated; spots were visualized under UV radiation and with I₂ vapour. Homogeneity of compounds was established by TLC and MS.

Extraction. Dried milled leaves and twigs of M. liliflora (1.2 kg) collected from Darjeeling, West Bengal during June 1979, were extracted exhaustively in a Soxhlet apparatus with hot petrol (60-80°) followed by CHCl₃ (40 hr each). The petrol extract on concn deposited a solid (fraction A) which was filtered. The filtrate (fraction B), residue (fraction A) and the neutral CHCl₃ portion (fraction C) obtained after separation of the basic constituents were separately chromatographed over Si gel (60-120 mesh) using solvents and solvent mixtures of increasing polarities. Compounds with the same R_f s (TLC) were combined. The CHCl₃ extract (positive Dragendorff's test) was concd under red. pres. to a thick gummy mass. The latter was extracted repeatedly by stirring with 5% citric acid soln (400 ml × 5). The combined acid soln (single spot in TLC) was carefully basified with NH_3 (pH ≥ 8) and the liberated base was taken up in CH_2Cl_2 (500 ml × 4). Evaporation of the solvent from the dried extract (dry Na₂SO₄) afforded a gummy residue which was chromatographed over neutral Al₂O₃ (Brockmann grade, BDH). The known compounds were characterized by spectroscopic methods (IR, UV, ¹H NMR, MS).

Isolation of (+)-veraguensin (3). The deep green residue from the C₆H₆-CHCl₃ (1:1) eluates of the main chromatography was subjected to repeated chromatography over Si gel. The later C₆H₆-CHCl₃ (1:1) eluates afforded (+)veragyensin (3) [9, 14, 17]: colourless flakes (40 mg, CHCl₃petrol), mp 125°, $[\alpha]_D = +31.83^\circ$ (CHCl₃; c 0.1037), R_f 0.3 in CHCl₃ (Si gel); pink colouration with conc. H₂SO₄; UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 232 (4.26), 279 (3.79): IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹; 3120, 2940 2880, 1600 (aromatic), 1585, 1502, 1225 (br), 1155, 1015 (br), 805, 750; 'H NMR (60 MHz, CDCl₃, TMS as int. standard): δ 3.90-3.95 (4 aromatic-OCH₃), 6.80-7.15 (6H, aromatic, br), 0.69 (3H, d, J = 6.3 Hz, H_3 C-3), 1.08 (3H, d, $J = 6 \text{ Hz}, \text{ H}_3\text{C}-4), 1.8-2.3 \text{ (2H, } m, \text{ H}-3 \text{ and H}-4), 4.45 \text{ (1H, } d,$ J = 8.3 Hz, H-5), 5.12 (1H, d, J = 8 Hz, H-2); MS 70 eV m/z(rel. int.): 372 [M]⁺ (26.1), 206 (100), 191 (35.8), 175 (31.6), 158 (10.3).

Isolation of futoenone (2). The black gummy residue obtained from the earlier CHCl₃-MeOH (19:1) eluates having two spots on TLC, was repeatedly chromatographed over Si gel. The early CHCl₃-MeOH (49:1) eluates afforded futoenone (2): fine colourless needles (55 mg, CHCl₃-petrol), mp 192°, R_f 0.4 in CHCl₃-MeOH (97:3) (Si gel), $[\alpha]_D = -60.0^\circ$ (CHCl₃: c 0.103), deep pink colouration with conc. H₂SO₄: UV $\lambda_{\max}^{\text{EIOH}}$ nm (log ϵ): 258 (4.26), 286 (3.95) IR ν_{\max}^{KBr} cm⁻¹: 2920(w), 1635 (conjugated C=O), 1600 (aromatic), 1480, 1235(br), 1030, 925 (-OCH₂O-), 725; MS 70 eV: m/z (rel. int.): 340 [M]⁺ (100), 178 (31.3), 177 (33.4), 176 (25.5), 164 (43.6), 163 (95.6), 162 (54.7), 150 (6.3), 149 (17.3), 148 (15.3), 147 (25.5), 135 (62.6 and 134 (17.7).

Isolation of (-)-maglifloenone (1). The mother liquor left after the removal of futoenone, and later CHCl₃-MeOH (49:1) eluates having another spot on TLC were subjected to rechromatography over Si gel. The later CHCl₃-MeOH (49:1) eluates afforded (-)-maglifloenone (1): colourless needles (25 mg, CHCl₃-petrol), mp 232°, R_f 0.35 in CHCl₃-MeOH (97:3), $[\alpha]_D = -90.12^\circ$ (CHCl₃; c 0.081), deep pink colouration with conc. H₂SO₄; UV λ_{\max}^{EOH} nm (log ε): 258 (4.32) and 285 (inflection); IR ν_{\max}^{KBr} cm⁻¹: 2938 (br), 1658 (conjugated C=O), 1608, 1590 (aromatic), 1506, 1240 (br), 1130 (br), 1000, 832; MS 70 eV m/z (rel. int.): 386 [M]⁺ (100), 222 (41.8), 210 (53.7), 209 (54.7), 208 (26.4), 207 (17.3), 196 (38.8), 195 (18.6), 182 (29.1), 181 (16.3), 179 (19.2), 177 (20.9), 164 (23.4) and 163 (16.5).

Isolation of taspine (4). Elution of the combined basic fractions with C_6H_6 -CHCl₃ (1:1) over neutral Al_2O_3

afforded taspine (4) which was further purified from CH₂Cl₂-petrol as an amorphous base (3 g), mp 225°d, R_f 0.6 in CHCl₃ (Al₂O₃ neutral), $[\alpha]_D = \pm 0^\circ$, no colouration with conc. H₂SO₄ and alcoholic KOH; UV λ_{max}^{EOH} nm (log ϵ): 247 (4.85), 285 (4.1) 333 (3.99) and 349 (4.07); UV $\lambda_{max}^{EOH+NaOH}$ nm (log ϵ): 215 (4.51) and 235 (4.55); IR ν_{max}^{KBr} cm⁻¹: 2950(w), 1725 (Σ C=O), 1598 (aromatic), 1460, 1435, 1282, 1182, 1085 and 970; Σ H NMR (80 MHz, CDCl₃, TMS as int. standard): Σ 2.39 [6H, s, -N(CH₃)₂], 2.66 (2H, m, H₂-5), 3.51 (2H, m, H₂-4), 4.10 (6H, s, H₃CO-2 and H₃CO-10), 7.19 (1H, s, H-3), 7.32 (1H, d, J = 9 Hz, H-9) and 8.21 (1H, d, J = 9 Hz, H-8); MS 70 eV m/z (rel. int.): 369 [M]⁺ (1) and 58 [CH₂=NMe₂]⁺ (100).

Taspine hydrochloride. A solution of taspine (50 mg) in CH_2Cl_2 containing a drop of conc. HCl (AR grade) was stirred for 2 hr and then cooled in ice. The separated solid crystallized from EtOH as colourless needles (7 mg), mp 251-252°d [18]. Its UV spectra [λ_{max}^{EtOH} nm (log ϵ): 248 (4.86), 285 (4.10), 333 (3.4) and 348 (4.08)] was similar to that reported in the literature [19].

Isolation of β -sitosterol. The earlier C_6H_6 fractions were combined and on chromatography over Si gel, C_6H_6 eluted β -sitosterol: flakes (50 mg, CHCl₃-MeOH), mp 138°, $[\alpha]_D = -37^\circ$; β -sitosterol acetate, mp 132°.

Acknowledgements—We express our thanks to Mr. D. F. Dance (Stirling University, U.K.) for MS measurements, to Mr. A. Acharya (this Department) and Dr. B. Ranu (IACS, Cal-32) for ¹H NMR measurements. The financial assistance of the UGC, New Delhi to P.C. by way of a JRF is gratefully acknowledged.

REFERENCES

 (1962) The Wealth of India: A Dictionary of Indian Raw Materials and Industrial Products, Raw Materials Vol. VI, p. 223. CSIR, New Delhi.

- Watanabe, K., Goto, Y. and Yoshitomi, K. (1973) Chem. Pharm. Bull. 21, 1700.
- 3. Fujita, M., Itokawa, H. and Sashida, Y. (1974) Yakugaku Zasshi 94, 729.
- 4. Ito, K. and Asai, S. (1974) Yakugaku Zasshi 94, 729.
- 5. Nakano, T. (1953) Pharm. Bull. 1, 29.
- 6. Nakano, T. (1956) Pharm. Bull. 4, 67.
- Tomita, M. and Nakano, T. (1952) J. Pharm. Soc. 72, 1260.
- 8. Talapatra, S. K., Chaudhuri, P. K. and Talapatra, B. (1980) Planta Med. 39, 222.
- Doskotch, R. W. and Flom, M. S. (1972) Tetrahedron 28, 4711.
- Ogiso, A., Kurabayashi, M., Mishima, H. and Woods, M. C. (1968) Tetrahedron Letters 2003.
- Woods, M. C., Miura, I., Ogiso, A., Kurabayashi, M. and Mishima, M. (1968) Tetrahedron Letters 2009.
- 12. Ogiso, A., Kurabayashi, M. and Takahashi, S. (1970) Chem. Pharm. Bull. 18, 105.
- Silverstein, R. M., Bassler, G. C. and Morril, T. C. (1974) in Spectrometric Identification of Organic Compounds, 3rd edn, p. 245. John Wiley & Sons, New York.
- Crossley, N. S. and Djerassi, C. (1962) J. Chem. Soc. 1459.
- Platonova, T. F., Kusowkow, A. D. and Scheinker, Yu. N. (1956) Zh. Obshch. Khim 26, 2651.
- 16. Safronich, L. N. (1961) Chem. Abstr. 55, 18892.
- McAlpine, J. B., Riggs, N. V. and Gordon, P. G. (1968) Aust. J. Chem. 21, 2095.
- Shamma, M. and Moniot, J. L. (1971) Chem. Commun. 1065.
- Holubek, J. and Strouf, O. (1966) Spectral Data and Physical Constants of Alkaloids. Spectrum No. 393.
 Publishing House of Czechoslovak Academy of Science, Prague.